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(19) (CA) APPLICATION FOR CANADIAN PATENT (12)

(54) Ibuprofen Triturates and Topical Compositions Containing  
Same

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**(54) Title: IBUPROFEN TRITURATES AND TOPICAL COMPOSITIONS CONTAINING SAME**

**(57) Abstract**

Ibuprofen forms a co-solution mixture which can be admixed with a vehicle to form a stable topical composition. Part of the menthol can be replaced by benzyl alcohol and the mixture can comprise also a pharmacologically acceptable alcohol, especially propylene glycol.

(Case 1)

IBUPROFEN TRITURATES AND TOPICAL COMPOSITIONS  
CONTAINING SAME

5 The present invention relates to topical ibuprofen formulations.

Ibuprofen (ie. (isobutyl phenyl)propionic acid) is a white crystalline drug, insoluble in water but relatively soluble in organic solvents such as alcohol (1 in 1.5);  
10 chloroform (1 in 1); ether (1 in 2) and acetone (1 in 1.5).

Ibuprofen usually is taken orally and, although a number of topical formulations have been proposed, we are not aware of any satisfactory topical formulations.

15 However, it often would be preferred to administer ibuprofen by topical application to an affected area so as to permit absorption through the skin. For example, in the treatment of rheumatic pain and/or inflammation, it is desirable to sustain a high local concentration of  
20 ibuprofen in the affected area of the body. Whereas oral administration to provide such local concentration would result in unacceptably high concentrations of ibuprofen throughout the body, topical application allows ibuprofen to accumulate only where it is needed.

25 In this connection, it is believed that solution dosage forms offer the best prospects of efficient percutaneous absorption of ibuprofen from topical formulations.

30 The present Applicants have experimented with topical formulations using conventional vehicles containing long chain cetyl and stearyl alcohols. They found that ibuprofen appears to react with these alcohols, thus reducing the concentration of the active ingredient for absorption. There was also a marked tendency for the  
35 ibuprofen to crystallise on storage as described above. Mineral and vegetable oils dissolve ibuprofen but,

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even at very high oil concentrations, the ibuprofen soon crystallises out. Oleic acid also dissolves ibuprofen but the solution has an unpleasant smell and is sticky and liable to autoxidation.

5 Vehicles based on combinations of oleic acid and medium chain length oils were investigated with various emulsifying agents - but avoiding cetyl and stearyl alcohols. These combinations prevented ibuprofen from crystallising out of an organic solution, but the emulsion formed could not be stiffened or settled sufficiently to withstand storage at 35 to 37°C for several months. Further, the H.L.B. (ie. hydrophilic-lipophilic balance) was higher than considered desirable for topical use.

10 Menthol (ie. 2-isopropyl-5-methylcyclohexanol) is a crystalline, naturally-occurring substance which has been used in pharmacy for at least a century. It has a penetrating odour and, for that reason, is widely used to relieve symptoms of bronchitis, sinusitis and similar conditions. It is used as an adjuvant in a number of 15 topical formulations and has been reported to enhance the percutaneous transfer of systemically active, water-soluble or solubilizable drugs (see EP-A-0147146).

20 It has long been known that trituration of menthol with certain other crystalline substances, such as camphor (ie. 1,7,7-trimethylbicyclo(2,2,1)heptan-2-one), chloral hydrate (ie. 2,2,2-trichloroethane-1,1-diol) and phenol, forms a co-solution liquid or soft mass. However, we are 25 not aware of any disclosure of trituration or admixture of menthol with a propionic acid derivative or of any other disclosure which would have led those skilled in the art 30 of topical formulations to consider the use of menthol to overcome the problem of topically formulating ibuprofen.

35 JP-A-63179820 discloses that the bioavailability of water-soluble pharmacological agents in suppository preparations can be increased by adding the agent to the suppository base as a solution in a mixture of menthol and

camphor. Ibuprofen is included in the list of specified agents.

It has now surprisingly been found that ibuprofen and menthol (both crystalline substances) form a co-solution when mixed together. Whilst not wishing to be bound to any particular theory, it is believed that the two substances form a co-solution or eutectic solution when so mixed. Depending upon the relative proportions of the two components, one or both may be present partially in microcrystalline form.

The use of such a co-solution in a topical pharmaceutical composition leads to improved stability. Further, co-solution mixtures of ibuprofen and menthol can be formed in situ by mixing ibuprofen and menthol together with other components of a topical composition.

It also has been found that the amount of menthol required to provide a co-solution with ibuprofen can be reduced by the presence of benzyl alcohol as a co-solvent for the ibuprofen.

The present invention provides a composition comprising a co-solution mixture of ibuprofen and menthol. The invention further provides a topical pharmaceutical composition containing said mixture in a pharmacologically acceptable vehicle and the use of menthol to stabilize a topical composition comprising ibuprofen.

A liquid triturate can be formed by triturating the ibuprofen, menthol and optionally other components at ambient temperature and the triturate added to a vehicle to form the topical pharmaceutical composition of the invention. However, the triturate could be formed by heating the components together whilst triturating or by stirring or otherwise mixing a fused mass of the components and then cooling the hot mixture.

Alternatively, the components can be compounded by dissolution in a suitable solvent, such as ethanol and the resultant solution (containing the co-solution mixture)

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added to the vehicle.

Any relative proportions of ibuprofen and menthol which provide a co-solution mixture which is liquid at ambient temperature can be used. Suitably, the amount of 5 ibuprofen by weight will be 10 to 70 percent based upon the weight of said mixture. However, said amount of ibuprofen preferably is 50 to 70 percent by weight.

Part of the menthol can be replaced by benzyl alcohol. When benzyl alcohol is present, it usually will 10 replace 10 to 80, preferably 25 to 60, weight percent of the menthol.

It is preferred to include a pharmaceutically acceptable alcohol, especially a glycol such as propylene glycol or a Macrogol (ie. polyethylene glycol) in the 15 liquid triturate of the invention when it is to be used to formulate a topical gel. When a triturate contains a glycol, it usually will be present in an amount by weight of up to 30 percent of the triturate. Preferably, said amount is 10 to 30 percent by weight and especially 20 to 20 25 percent by weight. However, substantially more glycol, eg. up to 70 percent by weight of the combined weight of ibuprofen, menthol, glycol and, if present, benzyl alcohol can be used when formulating a topical composition via a solution as mentioned above.

25 The liquid triturate or other co-solution mixture can be admixed with any compatible pharmacologically acceptable vehicle to form a topical composition.

Preferably, the vehicle is an aqueous gel or cream. 30 Carbomer (ie. carboxypolymethylene; carboxyvinyl polymer) is particularly suitable as a gelling agent in said aqueous gel vehicle.

Usually, the concentration of ibuprofen in the topical compositions of the invention will be in the range 0.1 to 20 percent by weight but any pharmacologically 35 active concentration can be used. Preferably, the ibuprofen concentration will be 2 to 12 percent by weight

and especially 3 to 6 percent by weight. Said amounts are by weight based upon the total weight of the topical composition.

5 The topical compositions of the invention can contain other compatible pharmacologically acceptable additives conventionally used in topical formulations such as antimicrobial agents, colorants, perfumes, Ph modifiers, antioxidants and stabilizers. Further, they can contain 10 other compatible pharmacologically active substances such as other non-steroidal anti-inflammatory agents, steroids, antibiotics and antibacterials.

The present invention is illustrated by the following non-limiting examples.

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EXAMPLE 1

4 g Crystalline ibuprofen was triturated with 4 g crystalline menthol to form an oil phase. This oil phase was then mixed with Carbomer, ethanol and water to form a 20 topical gel having the following composition:

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Ibuprofen	4 g
Menthol	4 g
Carbomer*	1-2 g
Ethanol	qs
Water	to 100 g

\* Carbopol 941

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EXAMPLE 2

4 g Crystalline ibuprofen was triturated with 4 g crystalline menthol and 5 g propylene glycol to form an oil phase, which was then mixed with Carbomer, ethanol and 35 water to form a topical gel having the following composition:

5            Ibuprofen            4 g  
          Menthol            4 g  
          Propylene glycol    5 g  
          Carbomer\*            1-2 g  
          Ethanol            qs  
          Water            to 100 g

10            \* Carbopol 941

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EXAMPLE 3

15            4 g Crystalline ibuprofen was triturated with 4 g  
          crystalline menthol, 3 g propylene glycol and 3 g benzyl  
          alcohol to form an oil phase, which was then mixed with  
          Carbomer, ethanol and water to form a topical gel having  
          the following composition.

20            Ibuprofen            4 g  
          Menthol            4 g  
          Propylene glycol    3 g  
          Benzyl alcohol      3 g  
          Carbomer\*            1-2 g  
25            Ethanol            qs  
          Water            to 100 g

\* Carbopol 941

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EXAMPLE 4

35            4 g Crystalline ibuprofen was triturated with 2 g  
          crystalline menthol, 4 g propylene glycol and 4 g benzyl  
          alcohol to form an oil phase, which was then mixed with  
          Carbomer, ethanol and water to form a topical gel having  
          the following composition.

Ibuprofen	4 g
Menthol	2 g
Propylene glycol	4 g
Benzyl alcohol	4 g
Carbomer*	1-2 g
Ethanol	qs
Water	to 100 g

10 \* Carbopol 941

**EXAMPLE 5**

3.0 g Crystalline ibuprofen was triturated with 1.5 g crystalline menthol to form an oil phase. This oil phase was then mixed with propylene glycol and then added to a second gel of Carbomer, ethanol and water to form a topical gel having the following composition:

20	Ibuprofen	3.0 g
	Menthol	1.5 g
	Propylene glycol	6.7 g
	Carbomer*	1-2 g
	Triethanolamine (85%)	1.25 g
25	Ethanol	23.0 g
	Water	to 100 g

\* Carbopol 980

30 The gel was cloudy in appearance and, under the microscope was seen to contain dispersed microcrystalline material.

**EXAMPLE 6**

35 Ibuprofen, menthol and propylene glycol were mixed

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together in ethanol and the resultant solution mixed with an aqueous Carbomer gel and subsequently thickened with triethanolamine to provide a topical gel of the same composition as that of Example 5.

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The gels of Examples 1 to 6 were found to be completely stable after storage for 6 months at ambient temperatures. At the end of the period of storage there was no significant loss of dissolved ibuprofen through 10 crystallisation.

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## CLAIMS:

1. A composition comprising a co-solution mixture of ibuprofen and pharmaceutical grade menthol.

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2. A topical pharmaceutical composition comprising a co-solution mixture of ibuprofen and pharmaceutical grade menthol in a pharmacologically acceptable vehicle.

10 3. A topical composition as claimed in Claim 2, wherein the vehicle is an aqueous gel.

15 4. A topical composition as claimed in Claim 2, wherein the ibuprofen content is 2 to 12 percent by weight of the composition.

20 5. A topical composition as claimed in Claim 4, wherein said ibuprofen content is 3 to 6 percent by weight of the composition.

6. A composition as claimed in Claim 1, which is a liquid triturate consisting essentially of ibuprofen and pharmaceutical grade menthol.

25 7. A composition as claimed in Claim 1, comprising benzyl alcohol.

8. A composition as claimed in Claim 7, which is a liquid triturate consisting essentially of ibuprofen, pharmaceutical grade menthol and benzyl alcohol.

30 9. A composition as claimed in Claim 1, wherein the ibuprofen is present in an amount in the range 50 to 70 percent by weight of the combined weights of ibuprofen, pharmaceutical grade menthol and, if present, benzyl alcohol.

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10. A composition as claimed in Claim 1, further comprising a glycol.

11. A composition as claimed in Claim 10, wherein said 5 glycol is propylene glycol.

12. A composition as claimed in Claim 11, which is a liquid triturate consisting essentially of ibuprofen, pharmaceutical grade menthol and propylene glycol.

10 13. A composition as claimed in Claim 10, wherein the glycol is present in an amount in the range 10 to 70 percent by weight of the combined weights of ibuprofen, pharmaceutical grade menthol, glycol and, if present, 15 benzyl alcohol.

14. A composition as claimed in Claim 13, wherein the amount by weight of glycol is 10 to 30 percent by weight of the triturate.

20 15. A composition as claimed in Claim 2, comprising ethanol.

25 16. A method of stabilizing a topical composition comprising ibuprofen which comprises incorporating in said composition an amount of crystalline menthol sufficient to form a co-solution mixture with the ibuprofen.

30 17. A topical composition as claimed in Claim 2, wherein said co-solution mixture is a liquid triturate consisting essentially of ibuprofen and pharmaceutical grade menthol.

35 18. A topical composition as claimed in Claim 17, wherein said co-solution mixture is a liquid triturate consisting essentially of ibuprofen, pharmaceutical grade menthol and benzyl alcohol.

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19. A topical composition as claimed in Claim 17, wherein said co-solution mixture is a liquid triturate consisting essentially of ibuprofen, pharmaceutical grade menthol and propylene glycol.

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20. A method of preparing an ibuprofen solution which comprises forming a co-solution mixture of ibuprofen with menthol from crystalline ibuprofen and crystalline menthol.

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